

Supramolecular Chemistry of Macrocyclic Oligomers of Metal Chelating Units

(Institute of Pure and Applied Sciences, University of Tsukuba) ○Takashi Nakamura

Keywords: Supramolecular Chemistry; Macrocycles; Coordination Bonds; Molecular Recognition; Dynamic Covalent Bonds

Macrocyclic oligomers of metal chelating units serve as unique ligands to synthesize multinuclear complexes that assemble coordination sites in the cavity (**Fig. 1a**).¹⁻⁴⁾ Such metallomacrocycles can realize precise molecular binding by utilizing multi-point coordination and rigidity of the metal complex units. Desymmetrization of macrocycles is another approach to create functional molecules, because they can employ their unsymmetrically arranged interaction moieties (**Fig. 1b**).^{1-3,5)}

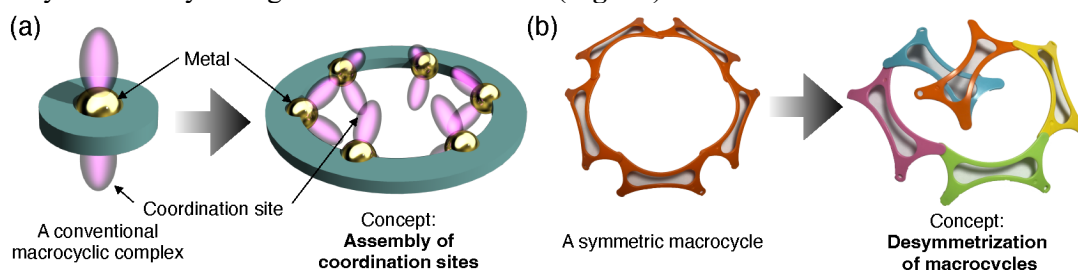


Figure 1. (a) Assembly of coordination sites. (b) Desymmetrization of macrocycles.

Pyridylmethylene-aminophenol (pap) hexamer and its Pd complexes as a metallohost^{2d)}

We have developed multinuclear complexes of hexapap, macrocyclic ligands with six pyridylmethylene-aminophenol (pap) units.²⁾ A hexanuclear Pd^{II} complex of hexapap [1Pd₆L₆](OTf)₆ (L: exchangeable ligand) has six inward coordination sites. Because of the planarity of one monomeric [Pd(pap)L] unit, [1Pd₆L₆](OTf)₆ can take two conformations. One is an *Alternate* conformation, in which six coordination sites of pap alternatively point to *Up-Down-Up-Down-Up-Down*. The other is a *Twisted* conformation, in which the coordination sites direct *Up-Middle-Down-Up-Middle-Down* (**Fig. 2**).

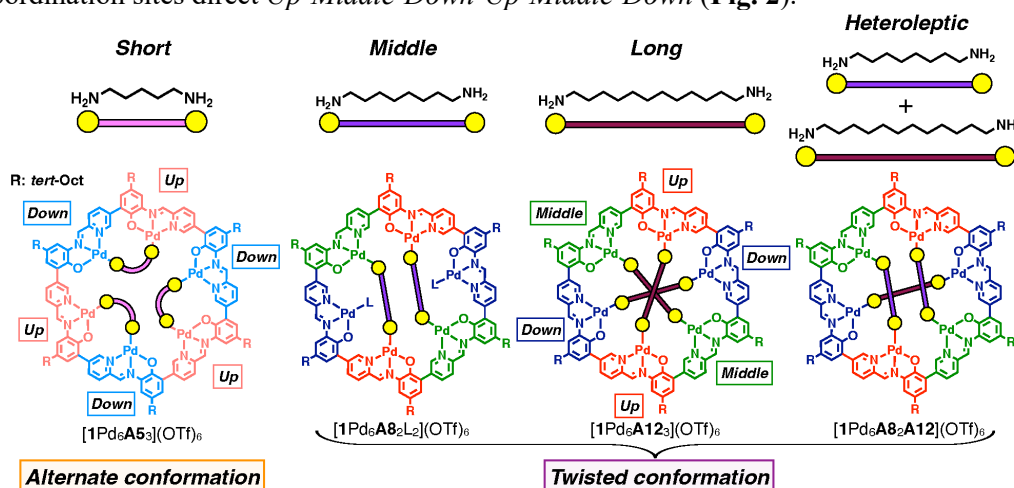


Figure 2. Site-selective ligand bridging of [1Pd₆L₆](OTf)₆ and its conformational regulation.

It was found that α,ω -diaminoalkanes $\text{H}_2\text{N}(\text{CH}_2)_n\text{NH}_2$ (**An**) bridged the inner coordination sites of $[\text{1Pd}_6\text{L}_6](\text{OTf})_6$ in different binding modes according to the number of methylene groups (n). **A4–A7** linked adjacent pap units to form $[\text{1Pd}_6\text{An}_3](\text{OTf})_6$ ($n = 4–7$) with *Alternate* conformation. **A8** connected 2 next pap units to yield $[\text{1Pd}_6\text{A8}_2\text{L}_2](\text{OTf})_6$ with *Twisted* conformation. **A12** connected diagonal Pd centers to result in $[\text{1Pd}_6\text{A12}_3](\text{OTf})_6$, again with *Twisted* conformation. Furthermore, in the case of **A8:A12** = 2:1, heteroleptic bridging was achieved to give $[\text{1Pd}_6\text{A8}_2\text{A12}](\text{OTf})_6$ (**Fig. 2**).

Pyridylbenzoxazole (pbo) trimer and its unsymmetric conversion³⁾

Synthesis of macrocycles with unsymmetric frameworks often suffers from multi-step reactions and low yields to sequentially connect different units. The use of dynamic covalent bonds is usually not effective for the synthesis of an unsymmetric macrocycle as a single product. We have achieved a high-yield 3-step synthesis of an unsymmetric macrocycle composed only of irreversible bonds (**Fig. 3**).

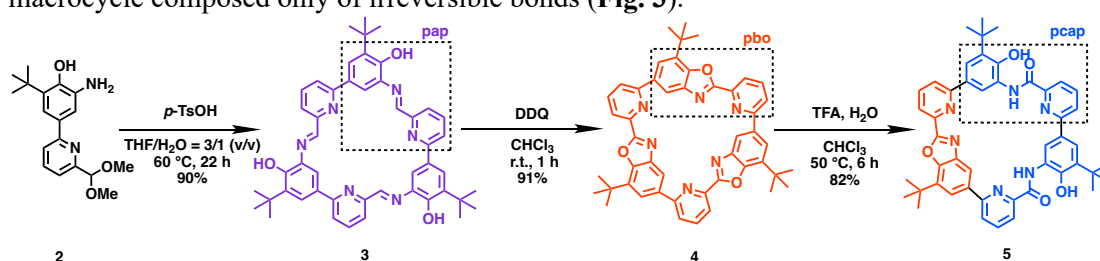


Figure 3. Synthesis of pcap-pbo mixed macrocycle **5** utilizing dynamic covalent bond

The three steps are comprised of macrocyclization, oxidation, and unsymmetric conversion. The first step is the macrocyclic oligomerization of the bifunctional monomer **2**, which has *o*-aminophenol and 2-formylpyridine subunits, utilizing imine bonds formation to yield the pap macrocycle **3**. The second step is the oxidation of the pap units of **3** to benzoxazole (pbo) units to yield the macrocycle **4** composed only of irreversible bonds. The last step is the unsymmetric conversion, that is, the selective addition reaction of water to only two of three pbo units of **3** to convert them into pyridylcarboxamidephenol (pcap) units, to obtain the unsymmetric macrocyclic ligand **5**. Moreover, the unsymmetric macrocycle **5** forms an interesting 2:1 complex with Zn^{II}, which holds the metal ion like a pearl in bivalve shells (**Fig. 4**).

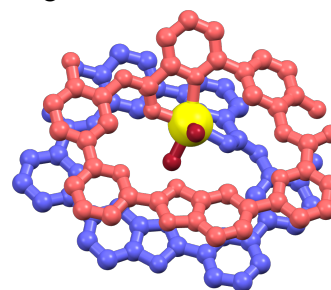


Figure 4. Structure of $[\text{5}_2\text{Zn}(\text{H}_2\text{O})_2]^{2+}$ determined by X-ray crystallography

- 1) T. Nakamura, *Chem. Lett.* **2021**, 50, 1822. (*Highlight Review*)
- 2) a) T. Nakamura, Y. Kaneko, E. Nishibori, T. Nabeshima, *Nat. Commun.* **2017**, 8, 129. b) A. Nagai, T. Nakamura, T. Nabeshima, *Chem. Commun.* **2019**, 55, 2421. c) T. Nakamura, R. Y. Feng, T. Nabeshima, *Eur. J. Inorg. Chem.* **2021**, 308. d) T. Nakamura, S. Watanabe, *Inorg. Chem.* **2023**, 62, 12886.
- 3) Y. Hokimoto, T. Nakamura, *Chem. Commun.* DOI: 10.1039/D3CC06216C
- 4) a) T. Nakamura, Y. Kawashima, E. Nishibori, T. Nabeshima, *Inorg. Chem.* **2019**, 58, 7863. b) T. Nakamura, S. Tsukuda, T. Nabeshima, *J. Am. Chem. Soc.* **2019**, 141, 6462.
- 5) a) T. Nakamura, S. Yonemura, T. Nabeshima, *Chem. Commun.* **2019**, 55, 3872. b) S. Yonemura, T. Nakamura, T. Nabeshima, *Chem. Lett.* **2020**, 49, 493. c) T. Nakamura, S. Yonemura, S. Akatsuka, T. Nabeshima, *Angew. Chem. Int. Ed.* **2021**, 60, 3080.