## Development of synthesis method for $\lambda\text{-Ti}_3O_5$ using titanium chloride as a starting material and evaluation of its heat-storage properties

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[Introduction] The lambda-phase trititanium pentoxide ( $\lambda$ -Ti<sub>3</sub>O<sub>5</sub>) was reported as a pressure-responsive heat-storage material that enable to preserve thermal energy in the long-term. <sup>[1], [2]</sup> In this study, we developed a synthesis method to obtain  $\lambda$ -Ti<sub>3</sub>O<sub>5</sub> simply by using titanium chloride as a starting material.

[Experiment] A mixed solution of  $H_2O$ ,  $TiCl_4$ , and  $NH_3$  was prepared in a round bottle flask. The solution was stirred at 50 °C for 20 hours in an oil bath. The precipitation was extracted from the solution by centrifugation, washed with ethanol, and heated at 60 °C for 24 hours to obtain a precursor. Then the precursor was sintered at 1100 °C for 20 hours under a hydrogen flow rate of  $0.5~\text{dm}^3~\text{min}^{-1}$ , obtained a black powder sample.

[Results] XRF measurement indicated that the composition formula was  $Ti_{3.00(3)}O_{5.00(3)}$  (Calculated: Ti 64.22, O 35.78 wt%; Found: Ti 64.53, O 35.47 wt%). Powder X-ray diffraction (PXRD) pattern and Rietveld analysis indicated that the obtained black powder was a single phase of  $\lambda$ -Ti<sub>3</sub>O<sub>5</sub> (monoclinic, C2/m; a = 9.8332(2), b = 3,78568(7), c = 9.9688(2),  $\beta = 91.259(2)$ °) and the crystalline size was estimated 57 ± 3 nm. Pressure threshold ( $P_{th}$ ) for converting 50 % of  $\lambda$ -Ti<sub>3</sub>O<sub>5</sub> to  $\beta$ -Ti<sub>3</sub>O<sub>5</sub> was approximately 300 MPa. The transition enthalpy ( $\Delta H_{trans}$ ) of the temperature-induced phase transition from pressure-produced  $\beta$ -Ti<sub>3</sub>O<sub>5</sub> to  $\lambda$ -Ti<sub>3</sub>O<sub>5</sub> was 7.78 ± 0.26 kJ mol<sup>-1</sup> at 462 K. Next, we investigate the influence of the crystalline size on the  $P_{th}$  and the  $\Delta H_{trans}$ . The examination of the relationship between crystalline size and the proportion of surface atoms intensify the influence of surface energy on the Gibbs free energy. This decreases the  $\Delta H_{trans}$  and consequently increases the  $P_{th}$ . Understanding the relationship between the crystalline size and heat storage properties is essential for developing effective heat storage materials.

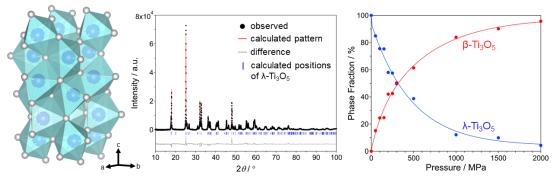


Fig. 1: Crystal structure of  $\lambda$ -Ti<sub>3</sub>O<sub>5</sub> (left), PXRD pattern with Rietveld analysis of the obtained black powder sample (middle), and the phase fraction of  $\lambda$ -Ti<sub>3</sub>O<sub>5</sub> and  $\beta$ -Ti<sub>3</sub>O<sub>5</sub> (right).

1) S. Ohkoshi, et al. Nature Chem., 2010, 2, 539. 2) H. Tokoro, et al., Nature Commun., 2015, 6, 7037.