Terbium single-molecule magnets embedded in an inorganic coordination framework showing self-thermometry using luminescence

(¹Institute of Pure and Applied Sciences, University of Tsukuba, ²Faculty of Chemistry Jagiellonian University, ³School of Science, The University of Tokyo)

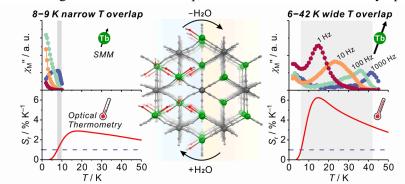
○Junhao Wang,¹ Jakub J. Zakrzewski,² Mikolaj Zychowicz,² Yue Xin,³ Hiroko Tokoro,¹

Shin-ichi Ohkoshi,³ Szymon Chorazy²

Keywords: Single-Molecule Magnet, Luminescence Thermometry, Emission

Research on combining magnetic and optical properties within a single-phase material unveils novel cross-effects, facilitating the development of advanced magneto-optical functional devices. Lanthanide(III) complexes are such materials, which can accommodate both luminescent functionalities and single-molecule magnet (SMM) property, promising for high-density information storage devices with optical features.

In this work, we constructed a three-dimensional cyanido-bridged coordination framework, [Tb^{III}(H₂O)₂][Co^{III}(CN)₆]·2.7H₂O (1), and its thermally desolvated phase, Tb^{III}[Co^{III}(CN)₆] (2), and investigated the conjunction of Tb(III)-centered SMM and luminescent functional properties in them. Upon topotactic transformation between two phases, the terbium complexes underwent a geometry change between an 8-coordinated square antiprism (in 1) and a 6-coordinated trigonal prism (in 2), which led to switching of both SMM and luminescent properties. Tb(III) in 1 revealed a moderate magnetic anisotropy with SMM character detectable up to 8.5 K. Upon dehydration, a much larger magnetic anisotropy energy barrier of 594(18) cm⁻¹ was detectable up to 42 K. Characteristic luminescence observed from Tb(III) exhibited a hue switching of in the emission color. Additionally, their emissions as a function of temperature made both 1 and 2 luminescent ratiometric thermometers, working in the cryogenic temperature regime as low as ca. 6 K, showcasing SMMs with self-temperature monitorable features by optical means.



1) Y. Xin, et al., J. Am. Chem. Soc. **2019**, 141, 18211. 2) J. Wang, et al., J. Am. Chem. Soc. **2020**, 142, 3970. 3) J. Wang, et al., Angew. Chem., Int. Ed. **2023**, 62, e202306372.