## Chirality Induction of Luminous Tb Complexes on Silica Surface Functionalized with Chiral Benzylamine Ligands

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**Keywords**: Terbium Complex; Chirality; Chiral Ligand; Silica Surface; Circularly Polarized Luminescence

Immobilization of chiral metal complexes on solid surfaces produces chiral metal complexes on solid materials, which are applied to chiral sensors and heterogeneous asymmetric catalysts etc. The preparation of new chiral molecular structures on  $SiO_2$  surface was investigated by the attachment of an achiral Tb complex on a  $SiO_2$  surface modified with chiral moieties and the chirality induction of an achiral Tb complex ( $^{Ph}1_{Tb}$ ) on a  $SiO_2$  surface with chiral benzylamine ligands ( $L_{NH2}(R/S)/SiO_2$ ) was successfully achieved.

A chiral ligand (L<sub>NHBoc</sub>(*R*/*S*)), whose amine moiety was protected with a *t*-butoxycarbonyl (Boc) group, was newly synthesized and characterized by <sup>1</sup>H NMR, FT-IR, and ESI-MS. L<sub>NHBoc</sub>(*R*/*S*) was attached to a SiO<sub>2</sub> surface, and the Boc group was finally deprotected to prepare L<sub>NH2</sub>(*R*/*S*)/SiO<sub>2</sub> (Figure 1). Solid-state CD spectra of L<sub>NH2</sub>(*R*/*S*)/SiO<sub>2</sub> showed inverted CD peaks derived from the surface-attached chiral benzylamine moieties, suggesting the formation of a homo-chiral SiO<sub>2</sub> surface with chiral benzylamine.

achiral Tb complex with tetraazacyclododecane-based tetrakis-(2-phenylphenoxide) ligand (Ph 1Tb) was immobilized on  $L_{NH2}(R/S)/SiO_2$  (Figure 1). Tb  $L_{III}$ -edge EXAFS of  $^{\rm Ph}1_{\rm Tb}/L_{\rm NH2}(R/S)/{\rm SiO}_2$  were similar to that of  $^{\rm Ph}1_{\rm Tb}$ , indicating that the coordination structure of the precursor  $^{Ph}1_{Tb}$  was maintained on the  $SiO_2$  surface (Figure 2). Solid-state CPL spectra of  $^{\text{Ph}}1_{\text{Tb}}/\text{L}_{\text{NH2}}(R/S)/\text{SiO}_2$  showed inverted CPL peaks derived from f-f emissions of Ph1Tb, strongly suggesting that the chirality of Ph1<sub>Tb</sub> was successfully induced on the attached SiO2 surface with the chiral benzylamine ligands.

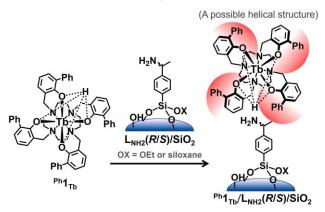
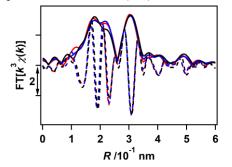


Figure 1. Preparation of Ph1<sub>Tb</sub>/L<sub>NH2</sub>(R/S)/SiO<sub>2</sub>.



**Figure 2.**  $k^3$ -weighted Tb  $L_{\rm III}$ -edge EXAFS Fourier transforms ( $k = 30 - 105 \text{ nm}^{-1}$ ) of  $^{\rm Ph}1_{\rm Tb}$  (black),  $^{\rm Ph}1_{\rm Tb}/L_{\rm NH2}(R)/{\rm SiO_2}$  (blue), and  $^{\rm Ph}1_{\rm Tb}/L_{\rm NH2}(S)/{\rm SiO_2}$  (red) (at 20 K). Solid lines: magnitude; dotted lines: imaginary parts.