Current Density Dependence of Discharge Termination Mechanisms in Lithium-Oxygen Secondary Batteries

(¹Graduate School of Engineering Science, Osaka University, ²Department of Materials Science and Engineering, Kyoto University, ³Institute of Multidisciplinary Research for Advanced Materials, Tohoku University, ⁴College of Science and Engineering, Tokyo Denki University, ⁵Research Center for Energy and Environmental Materials, NIMS)

○Terumi Goto,¹ Kiho Nishioka,¹,² Kazuyuki Iwase,¹,³ Yoshiharu Mukouyama,¹,⁴ Shuji Nakanishi¹,⁵

Keywords: Lithium Oxygen Batteries; Next-generation Secondary Batteries; Oxygen Reduction Reaction

Lithium-oxygen secondary batteries (Li-O₂ batteries) have attracted attention as a next-generation secondary battery due to their high theoretical gravimetric energy density. In this system, lithium peroxide (Li₂O₂), a solid with poor electrical conductivity, forms on the positive electrode during discharge. Ideally, this Li₂O₂ should be completely decomposed via oxidation at a low charging voltage; however, in reality, the charging voltage is high, and the cycle performance and energy efficiency of Li-O₂ batteries remain low. To address this issue, it is crucial to systematically understand the formation behavior of Li₂O₂ during the discharge process. In this study, we focused on the discharge termination mechanisms. Fig. 1 shows the discharge profiles at two different current densities. At a low current density (1 μ A cm⁻²), the discharge capacity reached 50 μ A h cm⁻². In this experiment, second discharge tests were conducted under the same conditions immediately after the first discharge termination. The second discharge at low current

density ceased almost immediately. According to the conventional understanding of the discharge determination in Li-O₂ batteries, this result is interpreted as being caused by the irreversible blockage of the cathode surface by Li₂O₂ deposition. On the other hand, at high current density (50 µA cm⁻²), the discharge capacity reached 5 µA h cm⁻². Notably, the first and second discharge curves at high current density overlapped, a phenomenon that be explained by the conventional understanding. This observation indicates that the cathode after discharge at high current density does not remain blocked by Li₂O₂ and is, in fact, reversible. These findings reveal two distinct discharge termination mechanisms dependent on the current density.

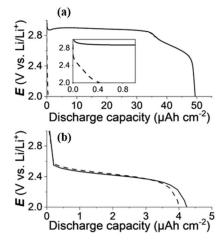


Fig. 1. Discharge profiles at (a)1 μ A cm⁻² and (b) 50 μ A cm⁻². The solid and dashed lines show the first and second discharge, respectively.