

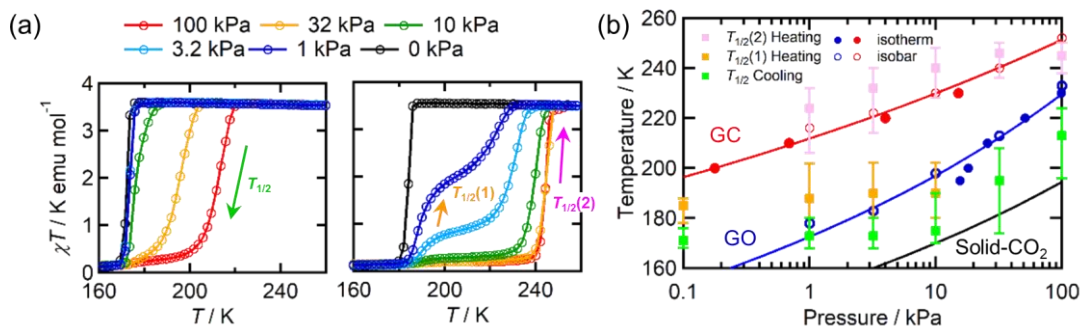
## CO<sub>2</sub>-actuated Control of Fe(II) Spin Crossover in a Hofmann-type Metal-Organic Framework

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The increased anthropogenic emission level of CO<sub>2</sub> urges the development of CO<sub>2</sub>-responsive materials, but is it possible to regulate the inherent electronic properties through weak physisorption of a ubiquitous gas such as CO<sub>2</sub>? Herein, we intended to answer this imperative question by the first case of CO<sub>2</sub>-actuated variable spin-state stabilization in an interdigitated Hofmann-type metal-organic framework [Fe<sup>II</sup>Pd(CN)<sub>4</sub>L<sub>2</sub>] (**1**, L = methyl isonicotinate) [1].

Compound **1** did not adsorb N<sub>2</sub> (77 K) and O<sub>2</sub> (90 K). In contrast, a sharp transition appeared in the CO<sub>2</sub> adsorption isotherm and isobar, indicating the occurrence of gated adsorption. Magnetic measurements were conducted using a home-built gas cell under an external magnetic field of 1 kOe and varied CO<sub>2</sub> pressure (Fig. 1a), showing a wide shift in transition temperature ( $T_{1/2}$ ) from 178 K at  $P_{\text{CO}_2}$  = 0 kPa to 229 K at  $P_{\text{CO}_2}$  = 100 kPa (defined by the average of cooling and heating). Interestingly, the emergence of a stepped behavior in the heating process below  $P_{\text{CO}_2}$  = 10 kPa and overlapping magnetic susceptibility values above  $P_{\text{CO}_2}$  = 10 kPa elucidate the selective low-spin state stabilization correlated with the extent of CO<sub>2</sub> accommodation. Based on the magnetic response and phase transition diagrams obtained under respective  $P_{\text{CO}_2}$  (Fig. 1b), a plausible scenario of the spin-state switching can be interpreted as [**1**(ls) + **1'**(ls)] → [**1**(hs) + **1'**(ls)] → **1**(hs) at  $P_{\text{CO}_2}$  ≤ 10 kPa, **1'**(ls) → **1**(hs) at 10 kPa <  $P_{\text{CO}_2}$  ≤ 32 kPa, and **1'**(ls) → **1'**(hs) → **1**(hs) at  $P_{\text{CO}_2}$  = 100 kPa, where **1** and **1'** represent CO<sub>2</sub>-free and CO<sub>2</sub>-accommodated states, respectively. The cooperative CO<sub>2</sub> adsorption with spin transition based on the varied CO<sub>2</sub> pressure corroborates a novel case for developing CO<sub>2</sub>-responsive magnetic materials.



**Fig. 1** (a)  $\chi T$ - $T$  plots under CO<sub>2</sub> on cooling (right) and heating (left). (b)  $T$ - $P$  phase diagram  
1) A. Paul, W. Kosaka, B. Kumar, D. J. Mondal, H. Miyasaka, S. Konar, *Chem. Sci.* **2024**, *15*, 15610.