## Phase Separation of Epsilon Iron Oxide and Barium Ferrite using a Reverse-Micelle and Sol-Gel Combined Synthesis

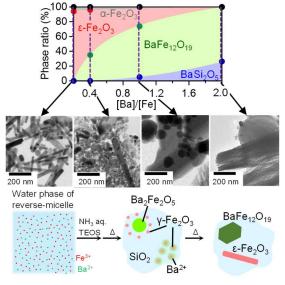
(¹Graduate School of Science, The University of Tokyo, ² Graduate School of Science, The University of Tsukuba) ○Jessica Grace MacDougall,¹ Hiroko Tokoro,² Marie Yoshikiyo,¹ Asuka Namai,¹ Shin-ichi Ohkoshi¹

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Hard magnetic ferrites, epsilon iron oxide ( $\epsilon$ -Fe<sub>2</sub>O<sub>3</sub>) and barium ferrite (BaFe<sub>12</sub>O<sub>19</sub>) are of significant interest due to their magnetic properties and potential applications in magnetic recording media and millimetre-wave absorption. This study investigates the effect of barium ion (Ba<sup>2+</sup>) to iron ion (Fe<sup>3+</sup>) ratio on the synthesis of  $\epsilon$ -Fe<sub>2</sub>O<sub>3</sub> and BaFe<sub>12</sub>O<sub>19</sub> using a combined reverse-micelle and sol-gel technique.<sup>1,2</sup> Reverse-micelle solutions were prepared containing Fe(NO<sub>3</sub>)<sub>3</sub> and varying concentrations of Ba(NO<sub>3</sub>)<sub>2</sub> (solution A), or ammonium hydroxide (solution B). Solution B was added dropwise into solution A, followed by tetraethyl orthosilicate and stirring, yielding a precipitate that was collected, annealed at 1000°C, and etched with NaOH (aq) and HCl (aq) to remove the silica matrix. By varying the Ba/Fe molar ratio, distinct phase separations and morphological changes were observed. At low Ba ion concentrations ([Ba]/[Fe] = 0.2),  $\epsilon$ -Fe<sub>2</sub>O<sub>3</sub> nanorods form predominantly, whereas higher Ba concentrations ([Ba]/[Fe]  $\geq$  0.4) favour the formation of BaFe<sub>12</sub>O<sub>19</sub>, shown by microscopy and powder X-ray diffraction (PXRD). At an equimolar ratio ([Ba]/[Fe] = 1),  $\epsilon$ -Fe<sub>2</sub>O<sub>3</sub> and BaFe<sub>12</sub>O<sub>19</sub> coexist in equal proportions, with no intermediate phases. Magnetic behaviour shows reduced coercive field and increased magnetic saturation as the major phase shifts from  $\epsilon$ -Fe<sub>2</sub>O<sub>3</sub> to

BaFe<sub>12</sub>O<sub>19</sub> as the Ba ion ratio is increased.

Based on the analysis conducted using scanning transmission electron microscopy coupled with energy dispersive X-ray spectroscopy and PXRD, the following mechanism is proposed for the sintering process: At 1000 °C, in areas of high barium ion concentration, γ-Fe<sub>2</sub>O<sub>3</sub> nanoparticles interact with Ba<sub>2</sub>Fe<sub>2</sub>O<sub>5</sub> to form BaFe<sub>12</sub>O<sub>19</sub>. Conversely, in areas of low barium ion concentration, γ-Fe<sub>2</sub>O<sub>3</sub> nanoparticles undergo transformation into ε-Fe<sub>2</sub>O<sub>3</sub> nanorods. This study clarifies the influence of the Ba/Fe ion ratio on this synthesis, facilitating the targeted



production of ferrite materials for advanced magnetic and technological applications.

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