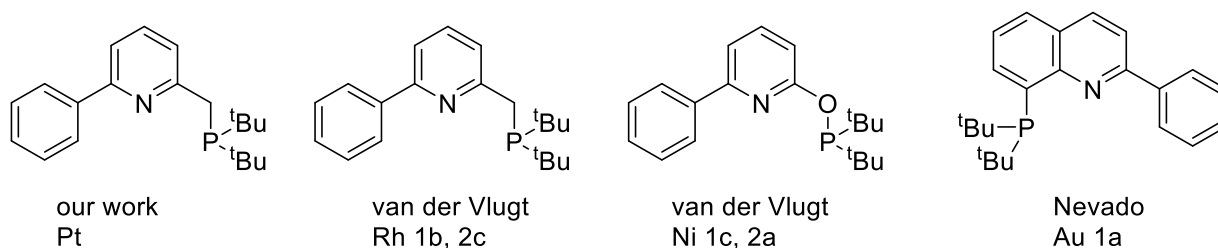


## The rare PNC pincer ligand, its Pt metalation, and reactivity of associated Pt complexes

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The 2phenyl-6bis-*t*-butylphosphinemethyl-pyridine (PNC) ligand and its close analogue, has only rarely been used in coordination chemistry<sup>1</sup> due to the difficulty of obtaining the desired pincer complexes cleanly, an inability to metalate the CH bond trans to the phosphine donor, and a tendency for facile cyclometallation of the pyridine ring instead of the pendant phenyl group. In addition, reductive elimination leading to a free phenyl moiety is a common decomposition pathway.<sup>2</sup>



In the referenced works, the ligand or its derivatives have been used for Au, Rh, and Ni complexation. However, the pincer ligand offers an opportunity to create highly reactive T-shaped Pt complexes by utilizing metal/ligand cooperation when dearomatizing the ligand with base. In the current presentation, we report on the synthesis of stable PNC-Pt pincer complexes and discuss our initial exploration of their reactivity. The initial PNC-Pt-Cl complex and its alkyl derivative are remarkably stable, even under air and at elevated temperatures. Attempts to deprotonate the ligand arm for metal/ligand cooperation reactivity in substrate activation will be discussed, as well as the reactivity and properties of the PNC-Pt-Me complex.

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