Ionic-Guests-Assisted Charge—Transfer Assemblies Based on the Electron-Donor Pillar[5] arene and the Electron-Acceptor Cyanostar

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Keywords: Pillar[5]arene; Cyanostar; Charge–Transfer Assembly; Co-crystallization; Electrostatic Interaction

Macrocycle-based charge—transfer assemblies (MCTAs) have gained increasing interest due to their potentials for guest recognition and the structural aesthetics. In most cases, MCTAs have been created by co-crystallization, using the rigid steric structures of macrocyclic molecules as the main backbone. However, when the sizes or shapes of the electron-donor and acceptor molecules are quite different, achieving close assembly of the molecules to effectively form MCTAs becomes significantly challenging.

Herein, we report the creation of MCTAs based on two macrocycles with different sizes and shapes; pillar[5] arene (PA) and cyanostar (CS). PA, composed of five alkoxy benzene units, provides an electron-rich cavity, thus exhibiting electron donor property (Fig. 1a). In contrast, CS, containing electron-withdrawing cyanostilbene units, displays electron acceptor properties (Fig. 1b). Therefore, by mixing PA and CS, charge—transfer assemblies can be formed. However, MCTAs obtained by mixing only PA and CS could not achieve effective CT emission. This would be because the structures of PA and CS differ significantly—PA has a 3D pillar-like shape and CS has a 2D planar shape (Fig. 1c). To address these issues, we focused on the fact that PA and CS are excellent receptors for cation and anion guest molecules, respectively, due to their electron-rich or electron-deficient cavities. By introducing ionic guest molecules into the PA—CS system, the cationic guests can be encapsulated by PA, while the anionic guests can be sandwiched by two CS (Fig. 1d). The electrostatic interaction between ionic guests allowed the PA and CS to assemble closely, resulting in an improvement of the quantum yield and enabling the observation of CT emission even in the solution state, which cannot be accessed without ionic guests.

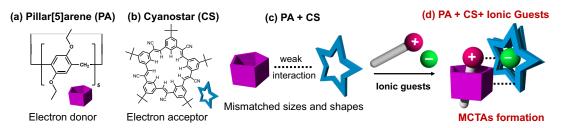


Fig.1 Chemical structures of (a) pillar[5] arene (PA) and (b) cyanostar (CS). CT assembly formation of (c) PA+CS and (d) PA+CS+ionic guests.

Reference: 1) J.-R. Wu, G. Wu, D. Li, Y.-W. Yang, Angew. Chem. Int. Ed. 2023, 62, e202218142.