## Transformable quadruply interpenetrated cage with multiple states of different reactivities

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Molecular self-assemblies with tunable multi-pockets are potentially functional molecules that generate multiple states. Pd<sub>4</sub>L<sub>8</sub> interpenetrated cages (ICs), in which two Pd<sub>2</sub>L<sub>4</sub> cages are quadruply interpenetrated, have three correlating, tunable pockets for anions. Here, we report that ICs binding three F or Cl ions, (F·F·F)- and (Cl·Cl·Cl)-ICs, were self-assembled from ditopic ligand L in high yields. The outer anions in the ICs were selectively exchanged with F or Cl under kinetic control, resulting in the quantitative formation of the thermodynamically most stable (Cl·F·Cl)-IC from the (F·F·F)-IC. The (F·Cl·F)-IC, which was produced from the (Cl·Cl·Cl)-IC with F by selective anion exchange, is a metastable transient species. The three thermodynamically stable ICs showed different reactivities in the presence of BF<sub>3</sub> and Ag<sup>+</sup>. The addition of BF<sub>3</sub> to the (Cl·Cl·Cl)-IC afforded heteroleptic [2]-catenane via unusual ligand exchanges, whereas the outer F<sup>-</sup> ions in the (F·F·F)-IC were removed by BF<sub>3</sub> to form a metastable ( $\otimes$ ·F· $\otimes$ )-IC. The heteroleptic [2]-catenane was converted into the (Cl·Cl·Cl)-IC by addition of Lewis base (F or Py\*). The addition of Ag to the (Cl·Cl)-IC caused dethreading to form the Pd<sub>2</sub>L<sub>4</sub> cages, whereas the (F·F·F)-IC was intact against Ag<sup>+</sup>. The thermodynamically most stable (Cl·F·Cl)-IC did not react with either BF<sub>3</sub> or Ag<sup>+</sup>. As a result, a total of seven states were generated, affording a complicated network in response to halide ions, Lewis acids and bases, and Ag<sup>+</sup>.

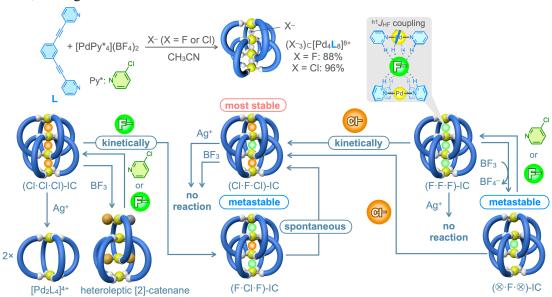


Figure Anion exchanges of quadruply interpenetrated cage (IC).

1) Tsukasa Abe, Yutong Zhang, Keisuke Takeuchi, and Shuichi Hiraoka, Chem, accepted.