

Reversible control of chirality and asymmetric charge transport behavior via an electrochemical approach

(¹*School of Science, Institute of Science Tokyo*) ○Po-Jung Huang,¹ Hiroaki Kusunoki,¹ Kouji Taniguchi¹

Keywords: Chirality; Transition Metal Dichalcogenides; Intercalation; Charge Transport

Chirality in materials opens up a wide variety of novel phenomena such as chirality-induced spin selectivity (CISS).¹ However, unlike charge and spin degrees of freedom, it is difficult to control chirality by an external field. This limits the manipulation of chirality-induced functionalities for further applications. Here, we demonstrate the electrochemical intercalation/deintercalation of chiral organic molecular cations (*R/S*-Mepy⁺) for a van der Waals 2D inorganic crystal of MoS₂ to reversibly control the chirality of the system. (Fig. 1a).

In the intercalated material, the shift of X-ray diffraction peaks to a lower 2θ angle was observed, indicating the expansion of interlayer distance. Phonon peaks in Raman spectra and Mo peaks in X-ray photoelectron spectra were both observed to shift to a lower energy. These imply the change of Mo oxidation state due to the electron doping. All these evidences support the successful electrochemical intercalation of *R/S*-Mepy⁺ into MoS₂. Importantly, the CISS is detected with a nearly full spin polarization and its sign reverses depending on the chirality of intercalated cation. Moreover, the intercalation process was reversible for at least 3 cycles (Fig. 1b). This work expands the practicable candidates from limited chiral materials to general 2D materials for the manipulation of chirality and the chirality-induced functionalities via the electrochemical approach.

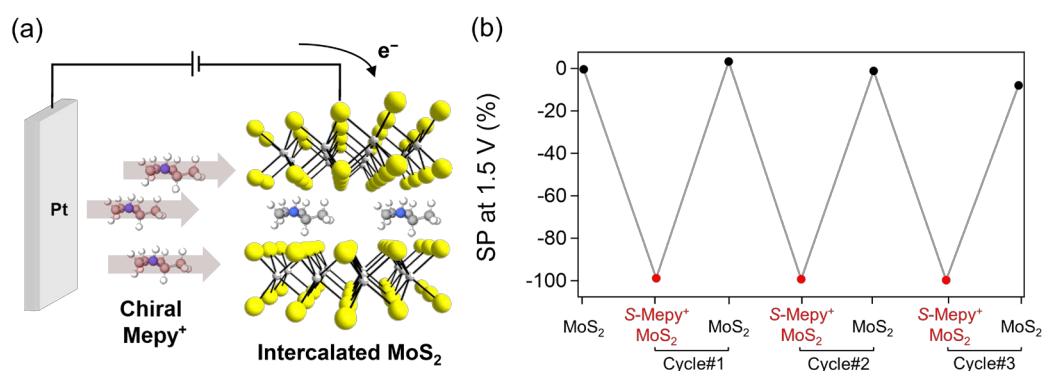


Fig. 1. (a) Schematic of the electrochemical intercalation of chiral molecular cations. (b) Electrochemical control of CISS with a nearly perfect spin polarization (SP).

1) B. P. Bloom, Y. Paltiel, R. Naaman, D. H. Waldeck, *Chem. Rev.* **2024**, 124, 1950.