

Coordination Polymers Constructed from Bis-pyridyl-bis-amide and 1,4,5,8-Naphthalenetetracarboxylic Acid: Ligand Transformation and Metal Ion Sensing

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Keywords: Coordination Polymer; Hydrothermal Reactions; Topology; Metal Ion Sensing;

Hydrothermal reactions of 1,4,5,8-naphthalenetetracarboxylic acid (1,4,5,8-H₄NTC) with various bis-pyridyl-bis-amide, *N,N'*-di(pyridin-3-yl)adipamide (**L**¹), *N,N'*-di(pyridin-3-yl)succinamide (**L**²), *N,N'*-di(pyridin-3-yl)oxalamide (**L**³) or *N,N'*-bis(pyridin-3-yl)methyl)oxalamide (**L**⁴) and transition metal salt afforded seven coordination polymers (CPs) and one dinuclear complex, including [Cd(**L**¹)_{0.5}(1,4,5,8-NTC')(H₂O)₂]_n (1,4,5,8-H₂NTC' = 1,4,5,8-naphthalenetetracarboxylic acid 1,8-monoanhydride), **1**, {[Zn(**L**²)_{0.5}(1,4,5,8-NTC')(H₂O)₂]·H₂O}_n, **2**, {[Cd(**L**²)_{0.5}(1,4,5,8-NTC')(H₂O)₂]·H₂O}_n, **3**, {[Zn(**L**⁵)(1,4,5,8-NTC')(H₂O)]·0.5OA}_n, **L**⁵ = 2,7-di(pyridin-3-yl)benzo[lmn][3,8]phenanthroline-1,3,6,8(2H,7H)-tetraone}, **4**, [Cd(**L**⁵)(1,4,5,8-NTC')(H₂O)]_n, **5a**, {[Cd(**L**⁵)(1,4,5,8-NTC')(H₂O)]·2H₂O}_n, **5b**, [Zn₂(**L**⁴)(1,4,5,8-NTC')₂(H₂O)₆], **6**, and {[Cd(**L**⁴)(1,4,5,8-NTC)_{0.5}(H₂O)]·0.5H₂O}_n (1,4,5,8-H₄NTC = 1,4,5,8-naphthalenetetracarboxylic acid), **7**. These CPs and dinuclear complex were structurally characterized by using X-ray single-crystal diffraction. CPs **1**, **2**, and **3** exhibit one-dimensional (1D) structures with the 2,3C2 topology and **4** and **5** display two-dimensional (2D) structures with the 2,4L1 topology, whereas **6** is a dinuclear complex and **7** shows a three-dimensional (3D) framework with the **mog** topology. Ligand transformations from 1,4,5,8-H₄NTC to 1,4,5,8-NTC'²⁻ and **L**³ to **L**⁵ are observed. The evaluation of the metal ion sensing properties of **1**, **2**, **3**, **6**, and **7** shows that **7** has the highest potential for detecting Fe³⁺ ions in aqueous solution with 1.06 × 10³ M⁻¹ and 1.99 × 10⁻⁴ M for the Stern-Volmer constant (*K*_{SV}) and the limit of detection (LOD), respectively.

