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Investigation of the Unique Co-catalytic Function of AlB₂ in Photocatalytic Overall Water Splitting

Effective co-catalysts are crucial for enhancing the activity of water splitting photocatalysts. While metals and transition metal oxides are commonly utilized as co-catalysts, we discovered that the intermetallic compound AlB₂ also functions as a cocatalyst. We demonstrated that AlB₂ powder improves the activity of the **KCaSrTa₅O₁₅ (KCSTO)** photocatalyst through simple mechanical mixing. In this study, we investigated how the synthesis conditions, composition, and loading amounts of AlB₂ affect its co-catalytic activity.

We synthesized **KCSTO** by solid-state reaction. AlB₂ was prepared by solid-state reaction of a mixture of elemental Al and B powders in evacuated sealed quartz tubes. For **Sample 1**, a mixture (Al:B = 1:2) was heated at 900 °C for 4 h. For **Sample 2**, a mixture (Al:B = 1.3:2) was heated at 900 °C for 3 h, followed by slow cooling from 900 °C to 800 °C over 1 h. These AlB₂ samples were then mechanically mixed with **KCSTO**. Photocatalytic activity was measured using an internal irradiation flow cell with a 400 W high-pressure Hg lamp.

According to X-ray diffraction patterns (**Fig. 1a**), **Sample 1** contained a considerable amount of AlB₁₂, with only a minor byproduct of α -Al₂O₃. For **Sample 2**, the formation of AlB₁₂ was almost suppressed. Photocatalytic measurements with these **AlB₂-KCSTO** composites (**Fig. 1b**) confirmed that overall water splitting occurred in all cases, exhibiting a stoichiometric H₂:O₂ ratio of 2:1. **Sample 1** showed ca. 3 times higher activity compared to pristine **KCSTO**. Notably, **Sample 2** achieved the highest photocatalytic activity. In conclusion, by adjustments of the raw material composition and temperature profiles during AlB₂ synthesis could prevent byproduct AlB₁₂ formation and consequently achieve a high co-catalytic performance.

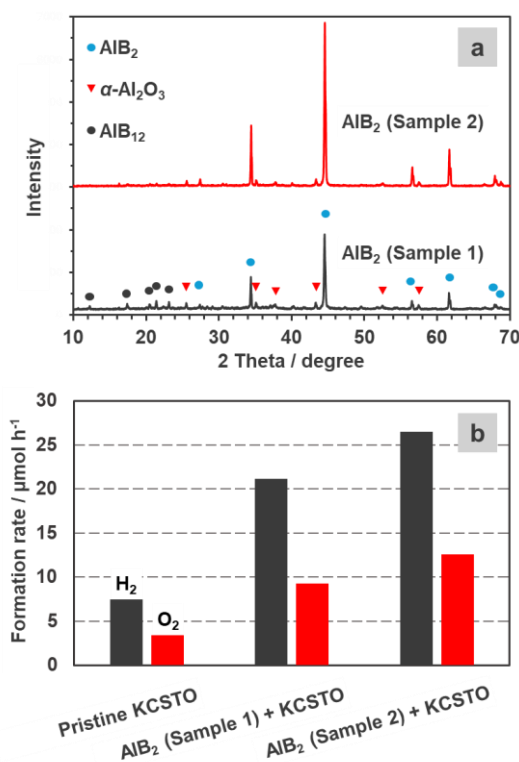


Fig. 1 XRD patterns of AlB₂ (a) and photocatalytic activity results using these samples (b).